

## Adsorption of oxygen molecules on individual single-wall carbon nanotubes

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Our study of the adsorption of oxygen molecules on individual semiconducting single-walled carbon nanotubes at ambient conditions reveals that the adsorption is physisorption, the resistance without O<sub>2</sub> increases by approximately two orders of magnitude as compared to that with O<sub>2</sub>, and the sensitive response is due to the pinning of the Fermi level near the top of the valence band of the tube, resulting from impurity states of O<sub>2</sub> appearing above the valence band. © 2006 American Institute of Physics. [DOI: 10.1063/1.2163008]

### I. INTRODUCTION

Interest in gas adsorption by carbon nanotubes at ambient conditions has been spurred by the demonstrations of the potential of single-walled carbon nanotube (SWCNT)-based gas sensors,<sup>1–3</sup> specifically, the establishment of the interdependence between gas adsorption and transport properties of carbon nanotubes (CNTs). In recent years, experimental studies on the adsorption of oxygen molecules by SWNT bundles or mats included the measurements of electrical resistance and thermoelectric power,<sup>2,3</sup> the effect of adsorption of O<sub>2</sub> on the barrier of metal-semiconductor contact,<sup>4,5</sup> and the kinetics of O<sub>2</sub> adsorption and desorption.<sup>6</sup> The picture emerged from these studies relevant to gas sensing indicates that the electrical resistance changes by about 15% between gassing and degassing,<sup>2</sup> that the hole doping of semiconducting single-walled nanotube (*s*-SWNT) in air is by the adsorption of O<sub>2</sub> in the bulk of *s*-SWNTs (Ref. 5) rather than at the contact,<sup>4</sup> and that the adsorption of O<sub>2</sub> has the characteristics of physisorption.<sup>6</sup> Theoretical investigations of the adsorption of O<sub>2</sub> on SWNTs have also been carried out, using spin-unpolarized as well as spin-polarized density-functional theory (DFT) methods.<sup>7–12</sup> Studies of the adsorption of O<sub>2</sub> on small-diameter (8,0) SWNT based on the spin-unpolarized DFT within the local-density approximation (LDA) predicted a weak hybridization between states of O<sub>2</sub> and those of the *s*-SWNT with an estimated charge transfer of  $\sim 0.1e$ ,<sup>7,9</sup> leading to a binding of O<sub>2</sub> at a distance less than 3 Å from the *s*-SWNT. The hole doping of the *s*-SWNT was attributed to the pinning of the Fermi level at the top of the valence band due to the adsorption of O<sub>2</sub>. With the O<sub>2</sub> molecule having a triplet ground state, the more realistic calculations based on the spin-polarized gradient-corrected DFT,<sup>8,10,12</sup> on the other hand, yielded a very weak bonding at  $\sim 4$  Å with no significant charge transfer, indicating that an O<sub>2</sub> molecule in the more stable triplet state is only physisorbed on a *s*-SWNT. For the triplet state of O<sub>2</sub> adsorbed on the (8,0) SWNT, two

degenerated  $pp\pi^*$  bands were found to split into four bands, with the two unoccupied  $pp\pi^*$  ( $\downarrow$ ) bands rising  $\sim 0.35$  eV above the top of the valence band at the  $\Gamma$  point,<sup>12</sup> casting some doubt about the hole-doping picture deduced from the unpolarized calculation.

In order to obtain a coherent and consistent picture of the adsorption of O<sub>2</sub> by individual *s*-SWNTs, we have conducted a careful experimental and theoretical investigation of the adsorption of O<sub>2</sub> molecule by individual SWNTs to shed light on the nature of adsorption and its effect on the transport properties of SWNTs. Experimentally, contacts were made to a few very dispersed SWNTs using *e*-beam lithography. The experiment was first conducted under ambient conditions in air (room temperature and atmosphere pressure). The resistance was monitored during each exposure to air and subsequent pumping ( $10^{-6}$  Torr). A resistance change of more than one order of magnitude was observed as a result of the adsorption of O<sub>2</sub> by *individual s*-SWNTs, in dramatic contrast to a mere 15% change observed for SWNT bundles or mats. Furthermore, the onset of the change in resistance was in minutes. These observations clearly demonstrated the feasibility of constructing *s*-SWNT-based chemical sensors. To be more consistent with the experimental result, we have carried out a study on the adsorption of an O<sub>2</sub> molecule by a larger SWNT than the one considered in previous studies, the (14,0) SWNT, that is closer to the range of diameters in the experiment, using the spin-polarized DFT method. Our calculation using the spin-polarized generalized gradient approximation (GGA) yielded a shallow potential well with depth of the order of  $\sim 0.05$  eV at  $\sim 3.6$  Å from the surface of the SWNT, consistent with the picture of physisorption. We have determined the pinning of the Fermi energy due to the impurity level associated with O<sub>2</sub>. Our estimate of the resistance based on the result of the (14,0) tube with the adsorption of the O<sub>2</sub> molecule is in excellent agreement with the observed initial resistance in air, indicating the metallization of the *s*-SWNT by hole doping associated with the physisorbed O<sub>2</sub>. We have also predicted a change in the resistance about two orders of magnitude between gassing and degassing.

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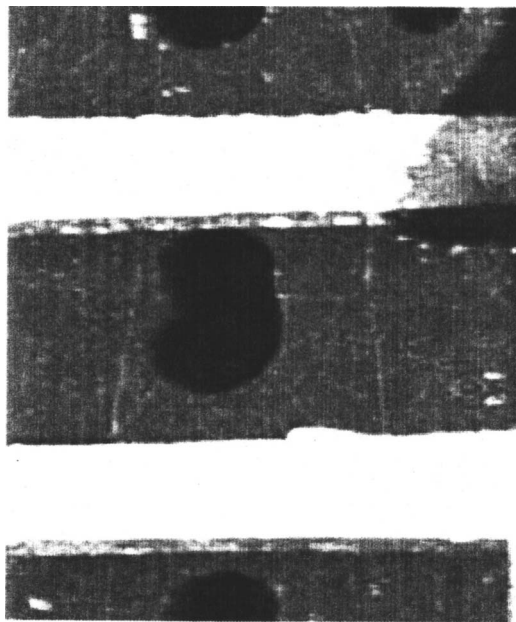


FIG. 1. An AFM image of two SWNTs buried underneath Au/Ti contacts.

## II. EXPERIMENTAL RESULTS

Individual SWNTs were synthesized using chemical vapor deposition (CVD) with an Fe catalyst and  $\text{CH}_4$  on a  $\text{SiO}_2/\text{Si}$  substrate with prepatterned grid marks. Silicon (100) with a thin oxide layer ( $0.4 \mu\text{m}$ ) was selected for the growth process. The grid pattern (Au alignment marks) was fabricated on the  $\text{SiO}_2/\text{Si}$  substrate using *e*-beam lithography and etched using a basic oxide etch (BOE). The alignment marks were etched so that they can be seen in atomic force microscopy (AFM) imaging. The preparation of the catalyst solution follows the procedure given in Ref. 13. Fe nanoparticles were dispersed on the substrate from  $\text{Fe}(\text{NO}_3)_3$  propanol solution. After washing with hexane, the substrate was loaded into the CVD reactor and heated to  $900^\circ\text{C}$  in flowing  $\text{Ar}/\text{H}_2$  [100 SCCM (standard cubic centimeter per minute) of 10%  $\text{H}_2$  in Ar]. After reduction at  $900^\circ\text{C}$  for 10 min, methane, the carbon feed, was introduced at a rate of 400 SCCM for 2 min. The sample was cooled in argon. The SWNT samples were imaged using the AFM with reference to the alignment marks in the grid pattern and the Au/Ti contacts were made on the SWNTs using *e*-beam lithography and evaporation. Larger contact pads were deposited on the *e*-beam-defined contacts using optical lithography (see Fig. 1). The device was loaded into a quartz reactor equipped with a turbo-molecular pump capable of evacuating to  $10^{-7}$  Torr for *in situ* studies. The reactor has provisions for gases and chemical vapors.

The experiment was first conducted under ambient conditions (room temperature and atmosphere pressure). The two-probe resistance of the device was measured during the exposure to air and subsequent pumping at room temperature. The resistance was continuously monitored during each exposure and subsequent pumping. Figure 2 shows the time evolution of the two-probe resistance of the device during pumping and subsequent exposure. The data for two cycles are shown. The two-terminal resistance of the as-prepared

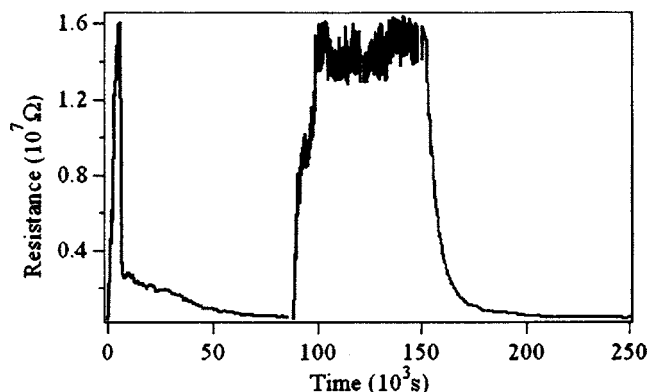


FIG. 2. Two cycles of the time evolution of the two-probe resistance of the sample during pumping and subsequent exposure to air.

device was  $\sim 300 \text{ k}\Omega$ . During pumping ( $\leq 10^{-6}$  Torr), the resistance started to increase and eventually saturated at a value of  $\sim 16 \text{ M}\Omega$  within a period of  $\sim 1 \text{ h}$ . This represents a change of the resistance of close to two orders of magnitudes for *individual* SWNTs, a dramatic change in comparison with the  $\sim 15\%$  change observed for SWNT bundles or mats.<sup>2</sup> When exposed to air at this point, the resistance started to decrease, initially with an abrupt drop to  $\sim 2.5 \text{ M}\Omega$  within  $\sim 15 \text{ min}$ . This substantial drop in the resistance within such a short time interval after the exposure to air indicates the sensitivity of the response of individual SWNTs to the absorption of gases in air. The initial drop in resistance was followed by a much slower decrease, saturating at the initial value of  $\sim 300 \text{ k}\Omega$  in  $\sim 10 \text{ h}$ . A similar behavior was observed in another cycle as shown in Fig. 2. The experimental findings suggest that the fabricated device is most likely composed of *s*-SWNTs and that the findings reflect the response of the transport properties of *s*-SWNT during the exposure to  $\text{O}_2$  in air and subsequent pumping. We established the semiconducting nature of our device by measuring the gate voltage dependence of the conductance of the device at room temperature, using a Si substrate as the back gate. We found that when the positive gate voltage is increased, the conductance decreases, while conductance increases when the negative gate voltage is increased. As the conductance of metallic tubes should have little or no gate voltage dependence, and on the other hand, an increasing negative gate voltage adds more holes to *p*-type *s*-SWNTs, thereby increasing the conductance, we conclude that our device consists of only *s*-SWNTs with the *p*-type behavior when exposed to air. This conclusion is consistent with the measurement of a positive thermoelectric power in the case of the adsorption of  $\text{O}_2$  molecules by SWNT bundles reported in Ref. 3, which indicates a *p*-type behavior for *s*-SWNTs with the adsorption of  $\text{O}_2$  molecules.

In Fig. 3 we show the gate voltage ( $V_g$ ) dependence of  $I_{\text{ds}}$  (for  $V_{\text{ds}}=300 \text{ mV}$ ) before and after the removal of air. The degenerately doped Si substrate was used as the back gate. The data were collected after the resistance reached the saturation values for both increasing and decreasing gate voltages. The  $I_{\text{ds}}$  vs  $V_g$  characteristics clearly show that the air-doped SWNTs (corresponding to the lowest resistance value) behave as a *p*-type semiconductor, i.e., they are on for a

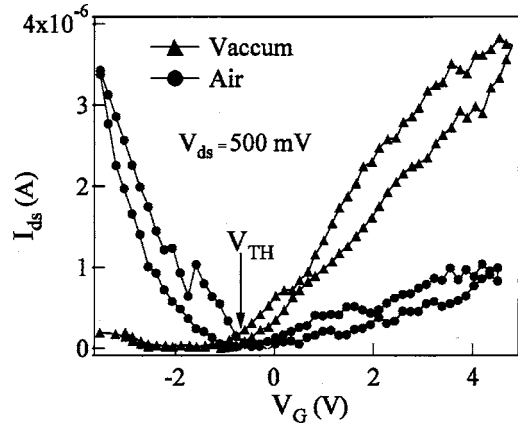


FIG. 3. Gate voltage dependence of  $I_{ds}$  before and after removal of air.  $V_{ds}=300$  mV. The data were collected after the resistance values reached the saturation values for both increasing and decreasing gate voltages.

negative gate bias. It shows some ambipolar properties as the device is not completely off for a positive gate bias, typical for most SWNT-based field-effect transistors (FETs). After pumping, its electronic character changes to  $n$  type (corresponding to the highest resistance value) as it exhibits a complete off state for a positive gate bias. The hysteresis observed in  $I_{ds}$  for decreasing and increasing gate voltages has been observed in most of the SWNT-based FETs and interpreted as due to trapped electrons. Most interestingly, the threshold voltage (marked in Fig. 3 by a downward arrow) for both  $p$ - and  $n$ -type devices is essentially the same. This is a conclusive indication that there is no charge transfer between the adsorbed  $O_2$  molecules and the  $s$ -SWNT, which is confirmed by our theoretical calculations, to be presented in Sec. III.

### III. THEORETICAL ANALYSIS

To shed light on the physics underlying the change in the transport properties during the adsorption and desorption of  $O_2$  molecules by  $s$ -SWNTs, we carried out a detailed study of the adsorption of an  $O_2$  molecule on a (14,0)  $s$ -SWNT, using spin-polarized LDA as well as spin-polarized GGA DFT methods in the Vienna *ab initio* simulation package (VASP).<sup>14–16</sup> We chose to use the (14,0) SWNT as the benchmark because its diameter ( $d=1.10$  nm) is close to the range of diameters of typical SWNTs and a recent DFT calculation has established the  $1/d$  dependence of the energy gap of  $s$ -SWNTs to be valid only for  $d \geq 1.0$  nm.<sup>17</sup> In our calcula-

tion, we used a supercell of size  $26 \times 26 \times 8.54$  Å to cut down the potential image effect between SWNTs. Along the axial direction of the SWNT, this supercell consists of two SWNT unit cells so that the calculation reflects well the situation of the physisorption of individual  $O_2$  molecules. Vanderbilt's ultrasoft pseudopotential<sup>18,19</sup> and the Perdew-Zunger functional,<sup>20</sup> with the GGA correction of Perdew *et al.*,<sup>21</sup> were used for the self-consistent spin-polarized solution. The energy cut off was set at 700 eV. Monkhorst-Pack scheme with  $1 \times 1 \times 11$   $k$ -point mesh was used for sampling the Brillouin zone. Full optimization of the structural configuration of SWNT+ $O_2$  and the lattice constants were carried out using the conjugate gradient method with the energy convergence of  $10^{-5}$  eV and forces  $\leq 10^{-2}$  eV/Å.

Our calculations, using spin-unrestricted LDA as well as GGA, confirmed that the triplet  $O_2$  state has a lower energy as compared with the singlet state. Before using the VASP code to investigate the benchmark case of the (14,0)  $s$ -SWNT+ $O_2$ , we applied in to the case of the (8,0)  $s$ -SWNT+ $O_2$  with  $O_2$  near the  $T$  site.<sup>12</sup> The optimization yields a result in excellent agreement with the corresponding result in Ref. 12 [see Fig. 5(g) in Ref. 12]. Having established the validity of the VASP code, we carried out optimizations of the adsorption of  $O_2$  on the (14,0)  $s$ -SWNT with spin-polarized methods (LDA and GGA). We found the binding to be the strongest for the triplet  $O_2$  molecule near the top of two adjacent zigzag bonds ( $T$  site), with the molecular axis perpendicular to the axial direction of the SWNT (see Fig. 4, left panel). The relaxed bonding geometries (bond length and equilibrium orientation of  $O_2$ ) from both methods are almost the same except for the equilibrium distance from  $O_2$  to the surfaces of the (14,0) SWNT. Figure 4 (right panel) shows a weak potential well of depth  $\sim 0.1$  eV at a distance of  $\sim 3.0$  Å for the LDA result and a very shallow well of  $\sim 0.03$  eV at a distance of  $\sim 3.5$  Å for the GGA result. These results are consistent with the scenario of physisorption. For physisorption characterized by weak interactions, LDA tends to overestimate the binding and underestimate the equilibrium distance, while GGA tends to underestimate the binding. From our result, one can conclude that the physisorption of  $O_2$  on the (14,0)  $s$ -SWNT is characterized by a potential well of depth between 0.03 and 0.1 eV and an equilibrium distance between 3.0 and 3.5 Å.

Figure 5 shows the band structures in the vicinity of the energy gap of relaxed configurations of the adsorption of triplet  $O_2$  on the surface of the (14,0)  $s$ -SWNT obtained by

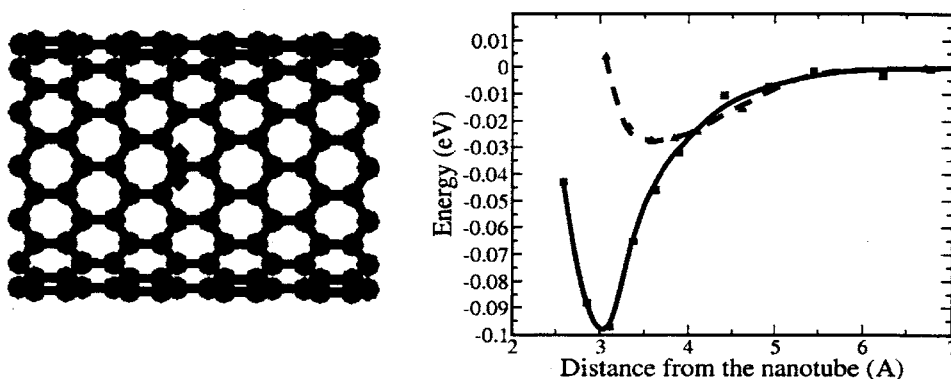


FIG. 4. The binding of a triplet  $O_2$  near the top of two adjacent zigzag bonds. Left panel: The equilibrium configuration. Right panel: Energy vs distance curves (dotted curve, GGA result; solid curve, LDA result).

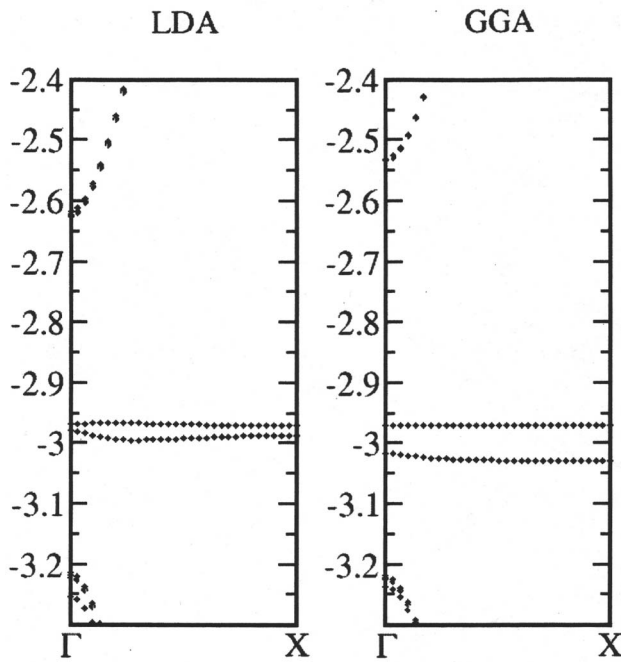


FIG. 5. Band structures corresponding to the equilibrium configurations. Left panel: LDA result; right panel: GGA result.

LDA and GGA, respectively. The energy gap obtained by the GGA calculation is  $\sim 0.69$  eV, while that by LDA is  $\sim 0.60$  eV. We have also checked the band structure for the pristine (14,0) *s*-SWNT using the same methods. We found the same values for the gap and no difference in the band structures as compared with those for the case of (14,0) +O<sub>2</sub> by the respective methods. Furthermore, the unoccupied oxygen  $pp\pi^*$ ( $\downarrow$ ) bands were found to appear within the gap of the *s*-SWNT almost dispersionless in both calculations. These results present an unambiguous indication of a very weak interaction between the oxygen molecule and the (14,0) *s*-SWNT, reinforcing the scenario of the physisorption. In this sense, our calculations have essentially established the placement of the empty impurity bands within the gap of the *s*-SWNT. Specifically, for the GGA calculation, the lower  $pp\pi^*$ ( $\downarrow$ ) band is  $\sim 0.20$  eV above the top of the valence band, while that for the LDA is  $\sim 0.24$  eV above the top of the valence band. To summarize, our study indicates no charge transfer between O<sub>2</sub> and the (14,0) *s*-SWNT. The effect of the presence of the oxygen impurity bands is to pin the Fermi level to the vicinity of the top of the valence band.

The conductance for the pristine *s*-SWNT and that for the *s*-SWNT with O<sub>2</sub> under ambient conditions can be estimated according to

$$G = \frac{2e^2}{h} \int_{-\infty}^{\infty} T(E) \left( -\frac{\partial f}{\partial E} \right) dE \approx G_0 \left( \frac{2}{1 + e^{\Delta/2kT}} \right), \quad (1)$$

where  $T(E)$  is the transmission coefficient as a function of  $E$  and may be approximated by 2 in the vicinity of the Fermi energy for SWNTs,  $f(E)$  is the Fermi distribution function,  $G_0 = 4e^2/h$  is the quantum conductance, and  $\Delta$  is the energy gap. Using Eq. (1) based on the energy gap with or without O<sub>2</sub> obtained by LDA as well as GGA, we have calculated the resistances of the (14,0) *s*-SWNT with or without O<sub>2</sub> at room temperature. The results are shown in Table I. It can be seen that the GGA method yields a value of  $\sim 400$  k $\Omega$  for the resistance with O<sub>2</sub>, in very good agreement with the experimental result, while the LDA method gives rise to a value of  $\sim 680$  k $\Omega$ . For the (14,0) *s*-SWNT, the GGA method leads to a resistance increase by a factor of  $\sim 1.33 \times 10^4$  between the resistance of the *s*-SWNT without O<sub>2</sub> and that with O<sub>2</sub>, while the LDA leads to an increase in resistance by a factor of  $1.08 \times 10^3$ . This resistance change can be attributed to the pinning of the Fermi level to the vicinity of the valence band due to the presence of the empty oxygen bands. Since the diameter of a typical SWNT is  $\sim 1.40$  nm and the gap follows a  $1/d$  dependence on the diameter for  $d \geq 1$  nm, we estimated the energy gap of the typical *s*-SWNT using the calculated gap of the (14,0) SWNT ( $d = 1.09$  nm) according to  $\Delta_{\text{adj}} = \Delta \times 1.09/1.4$ . Using  $\Delta_{\text{adj}}$ , we obtained a resistance increase by a factor of 82 for the LDA result and 652 for the GGA result, consistent with the experimental result.

#### IV. DISCUSSION

Our experimental study and theoretical analysis clearly lead to the following conclusion concerning the adsorption and desorption of O<sub>2</sub> molecules on *s*-SWNTs: (i) The resistance change between the desorption and absorption of O<sub>2</sub> molecules by an *individual s*-SWNT is approximately two orders of magnitude. The response of individual *s*-SWNTs to the exposure to O<sub>2</sub> molecules is therefore far more sensitive as compared with the response of SWNT bundles or mats studied previously. (ii) The adsorption of O<sub>2</sub> molecules on *s*-SWNTs is unequivocally physisorption. There is no charge transfer between the O<sub>2</sub> molecules and the *s*-SWNT. (iii) The sensitive response of *s*-SWNTs to the adsorption of O<sub>2</sub> molecules is due to the pinning of the Fermi level near the top of the valence band.

TABLE I. Resistances of the (14,0) SWNT without (w/o) and with O<sub>2</sub> at room temperature, and their ratio, calculated by spin-polarized LDA and GGA, respectively. Also shown are the adjusted resistances corresponding to a *s*-SWNT (without O<sub>2</sub>) with a diameter of 1.40 nm.

	$R$ w/o O <sub>2</sub> (k $\Omega$ )	$R$ with O <sub>2</sub> (k $\Omega$ )	Ratio	$R$ w/o O <sub>2</sub> (k $\Omega$ ) adjusted	Ratio adjusted
LDA	$7.4 \times 10^5$	$6.8 \times 10^2$	$1.08 \times 10^3$	$5.6 \times 10^4$	$8.20 \times 10^1$
GGA	$5.2 \times 10^6$	$4.0 \times 10^2$	$1.33 \times 10^4$	$2.6 \times 10^5$	$6.52 \times 10^2$

**ACKNOWLEDGMENTS**

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